

## Supplementary Information

# Unveiling the reactions of triethylphosphite and its diethylamino substituted derivatives to carbon tetrachloride with a molecular electron density theory perspective

Ali Barhoumi<sup>a</sup>, Mohammed Salah<sup>a</sup>, Abdellah Zeroual<sup>a</sup>, Mrinmoy Chakraborty<sup>b</sup> & Nivedita Acharjee<sup>c,\*</sup>

<sup>a</sup>Molecular Modeling and Spectroscopy Research Team, Department of Chemistry, Faculty of Sciences, Chouaib Doukkali University,  
El Jadida, Morocco

<sup>b</sup>Department of Electronics and Communication Engineering, Dr. B. C. Roy Engineering College, Durgapur-713 206, West Bengal, India

<sup>c</sup>Department of Chemistry, Durgapur Government College, Durgapur-713 214, West Bengal, India

\*E-mail: nivchem@gmail.com

Received 29 August 2021; revised and accepted 04 January 2022

S.No	Contents	Pg.No.
1	<b>Table S1.</b> B3LYP/6-311G(d,p) calculated ELF basin populations at threagents <b>1- 4</b>	<b>2</b>
2	<b>Table S2.</b> B3LYP/6-311G(d,p) calculated total energies, enthalpies, entropiesand Gibbs free energies, computed in au of the stationary points involved in the reactions of <b>1, 3 and 4</b> with <b>2</b> .	<b>3</b>
3	<b>Table S3.</b> B3LYP/6-311G(d,p) calculated ELF basin populations at <b>TS1C</b> and <b>TS1CI</b>	<b>4</b>

Table S1 — B3LYP/6-311G(d,p) calculated  
ELF basin populations at the reagents **1- 4**

	<b>1</b>	<b>3</b>	<b>4</b>	<b>2</b>
V(O1)	4.77			
V(O2)	2.38	2.36		
V'O2)	2.47	2.49		
V(O3)	2.19	2.31	2.47	
V'(O3)	2.69	2.59	2.41	
V(P,O1)	1.57			
V(P,O2)	1.46	1.45		
V(P,O3)	1.41	1.39	1.38	
V(N1)		1.69	1.74	
V(N2)			1.01	
V'(N2)			1.63	
V(N3)				
V(P,N1)		2.63	2.62	
V(P,N2)			1.7	
V(P,N3)				
V(P)	2.23	2.23	2.19	
V(C11)				2.17
V'(C11)				2.13
V''(C11)				2.09
V(C12)				2.13
V'(C12)				2.17
V''(C12)				2.09
V(C13)				2.09
V'(C13)				2.17
V''(C13)				2.13
V(C14)				2.09
V'(C14)				2.13
V''(C14)				2.17
V(C,C11)				1.55
V(C,C12)				1.51
V(C,C13)				1.53
V(C,C14)				1.48

Table S2 — B3LYP/6-311G(d,p) calculated total energies, enthalpies, entropies and Gibbs freeenergies, computed in au of the stationary points involved in the reactions of **1**, **3** and **4** with **2**.

	E	H	S	G
<b>1</b>	-804.897505	-804.669764	121.53	-804.727506
<b>2</b>	-1878.977053	-1878.961267	74.127	-1878.996487
<b>3</b>	-863.653825	-863.354744	137.351	-863.420004
<b>4</b>	-922.404373	-922.034934	141.799	-922.102307
<b>TS1C</b>	-2683.788096	-2683.543208	164.755	-2683.621488
<b>P1C</b>	-2683.91232	-2683.667663	196.573	-2683.761061
<b>TS1CI</b>	-2683.812938	-2683.568445	175.92	-2683.65203
<b>P1CI</b>	-2683.94679	-2683.701455	196.916	-2683.795016
<b>TS2C</b>	-2742.546197	-2742.229788	172.771	-2742.311877
<b>P2C</b>	-2742.673044	-2742.356724	209.057	-2742.456053
<b>TS2CI</b>	-2742.576436	-2742.260457	183.745	-2742.34776
<b>P2CI</b>	-2742.71107	-2742.39401	209.354	-2742.493481
<b>TS3C</b>	-2801.305607	-2800.917513	184.417	-2801.005136
<b>P3C</b>	-2801.420309	-2801.03231	217.69	-2801.135741
<b>TS3CI</b>	-2801.337874	-2800.95041	195.42	-2801.04326
<b>P3CI</b>	-2801.464515	-2801.075877	220.998	-2801.18088

Table S3 — B3LYP/6-311G(d,p) calculated most significant ELF basin populations at **TS1C** and **TS1Cl**

	<b>TS1C</b>		<b>TS1Cl</b>
V(C127)	0.13	V(C129)	2.50
V(C127)	2.49	V(C119)	1.22
V(C127)	0.11	V(C119)	4.56
V(C128)	1.22	V(C129)	1.50
V(C128)	1.52	V(C129)	2.43
V(C128)	2.44	V(C119)	2.44
V(C128)	2.04	V(C128)	2.45
V(C129)	0.08	V(C128)	1.91
V(C129)	2.41	V(C128)	2.47
V(C129)	0.08	V(C130)	2.16
V(C129)	0.19	V(C130)	2.47
V(C130)	1.99	V(C130)	1.35
V(C130)	0.51	V(C130)	1.36
V(C130)	0.64		
V(C130)	1.51		
V(C130)	2.33		
V(C130)	0.50		
V(C130)	1.50		
v(C26,C129)	0.43	V(C27,C130)	2.00
V(C26, C128)	1.72	V(C27,C128)	1.52
V(C26, C127)	1.72	V(C27,C129)	1.37
V(O2)	2.46	V(O2)	2.39
V(O2)	2.35	V(O2)	2.42
V(O3)	2.45	V(O3)	2.39
V(O3)	2.10	V(O3)	2.41
V(O4)	2.08	V(O4)	2.47
V(O4)	2.37	V(O4)	2.35
V(O4)	2.45		
V(P)	1.90	V(P1)	1.51
V(C26)	1.53	V(C27)	1.73
V(P1,O4)	2.12	V(P1,O4)	1.52
V(P1,O3)	1.73	V(P1,O3)	1.53
V(P1,O2)	1.52	V(P1,O2)	1.67